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APPLICATION NO.	FI	LING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/076,670	(02/19/2002	Roger R. Lesieur	C-2407 Cont. 2838	
7	590	05/05/2004		EXAMINER	
William W. J	ones		NGUYEN, TAM M		
Patent Counsel				ART UNIT	PAPER NUMBER
6 Juniper Lane Madison, CT				1764	
,				DATE MAILED: 05/05/2004	4

Please find below and/or attached an Office communication concerning this application or proceeding.

•	Application No.	Applicant(s)	1
	10/076,670	LESIEUR ET AL.	\mathcal{H}
Office Action Summary	Examiner	Art Unit	
	Tam M. Nguyen	1764	
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence addre	·ss
A SHORTENED STATUTORY PERIOD FOR REPLY THE MAILING DATE OF THIS COMMUNICATION. - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a reply If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	36(a). In no event, however, may a reply be time within the statutory minimum of thirty (30) days will apply and will expire SIX (6) MONTHS from a cause the application to become ABANDONEI	nely filed s will be considered timely. the mailing date of this comm D (35 U.S.C. § 133).	unication.
Status			
1) Responsive to communication(s) filed on 19 Fe	ebruary 2002.		
2a) ☐ This action is FINAL . 2b) ☑ This	action is non-final.		
3) Since this application is in condition for allowar			erits is
closed in accordance with the practice under E	x parte Quayle, 1935 C.D. 11, 45	33 O.G. 213.	
Disposition of Claims			
4)⊠ Claim(s) <u>1-12</u> is/are pending in the application.			
4a) Of the above claim(s) is/are withdraw	vn from consideration.	•	
5) Claim(s) is/are allowed.			
6)⊠ Claim(s) <u>1-12</u> is/are rejected.	•		
7) Claim(s) is/are objected to.			
8) Claim(s) are subject to restriction and/or	r election requirement.		
Application Papers			
9) The specification is objected to by the Examine	r.		•
10)⊠ The drawing(s) filed on <u>19 February 2002</u> is/are		d to by the Examiner.	•
Applicant may not request that any objection to the			
Replacement drawing sheet(s) including the correct	ion is required if the drawing(s) is obj	ected to. See 37 CFR	1.121(d).
11)☐ The oath or declaration is objected to by the Ex	aminer. Note the attached Office	Action or form PTO-	152.
Priority under 35 U.S.C. § 119			
12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of: 1. Certified copies of the priority documents 2. Certified copies of the priority documents 3. Copies of the certified copies of the priori	s have been received. s have been received in Application ity documents have been receive u (PCT Rule 17.2(a)).	on No ed in this National Sta	age
* See the attached detailed Office action for a list Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08)	4)	(PTO-413)	2)
3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) Paper No(s)/Mail Date	6) Other:		,

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DETAILED ACTION

Double Patenting

The nonstatutory double patenting rejection is based on a judicially created doctrine grounded in public policy (a policy reflected in the statute) so as to prevent the unjustified or improper timewise extension of the "right to exclude" granted by a patent and to prevent possible harassment by multiple assignees. See *In re Goodman*, 11 F.3d 1046, 29 USPQ2d 2010 (Fed. Cir. 1993); *In re Longi*, 759 F.2d 887, 225 USPQ 645 (Fed. Cir. 1985); *In re Van Ornum*, 686 F.2d 937, 214 USPQ 761 (CCPA 1982); *In re Vogel*, 422 F.2d 438, 164 USPQ 619 (CCPA 1970); and, *In re Thorington*, 418 F.2d 528, 163 USPQ 644 (CCPA 1969).

A timely filed terminal disclaimer in compliance with 37 CFR 1.321(c) may be used to overcome an actual or provisional rejection based on a nonstatutory double patenting ground provided the conflicting application or patent is shown to be commonly owned with this application. See 37 CFR 1.130(b).

Effective January 1, 1994, a registered attorney or agent of record may sign a terminal disclaimer. A terminal disclaimer signed by the assignee must fully comply with 37 CFR 3.73(b).

Claims 1-9 are rejected under the judicially created doctrine of obviousness-type double patenting as being unpatentable over claims 1 and 2 of U.S. Patent No. 6,533,924. Although the conflicting claims are not identical, they are not patentably distinct from each other because both sets of claims claim a desulfurizing process by contacting a fuel stream with a nickel reactant station.

The patented claimed set does not disclose that the fuel stream is a diesel fuel. However, it would have been obvious to one having ordinary skill in the art at the time the invention was made to have modified the patented claimed process by treating a diesel fuel because any hydrocarbon fuel which contains sulfur can be treated in the process. Therefore, it would be expected that diesel would be effective to treat in the process the patented claimed process.

Claims 1-9 are rejected under the judicially created doctrine of obviousness-type double patenting as being unpatentable over claim 1 of U.S. Patent No. 6,454,935. Although the conflicting claims are not identical, they are not patentably distinct from each other because both

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sets of claims claim a desulfurizing process by contacting a fuel stream with a nickel reactant station.

The patented claimed set does not disclose that the fuel stream is a diesel fuel. However, it would have been obvious to one having ordinary skill in the art at the time the invention was made to have modified the patented claimed process by treating a diesel fuel because any hydrocarbon fuel which contains sulfur can be treated in the process. Therefore, it would be expected that diesel would be effective to treat in the process the patented claimed process.

Claims 1-12 are provisionally rejected under the judicially created doctrine of obviousness-type double patenting as being unpatentable over claims 1, 2, and 4-11 of copending Application No. 10/042,056. Although the conflicting claims are not identical, they are not patentably distinct from each other because both sets of claims claim a desulfurizing process by contacting a fuel stream with a nickel reactant station.

The patented claimed set does not claim that the low sulfur content fuel is suitable for use in an internal combustion. However, the low sulfur content fuel of the patented claimed set is a hydrocarbon fuel. Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to have modified the patented claimed set by utilizing the low sulfur fuel in an internal combustion because a hydrocarbon fuel is suitable in any fueling system which is either a fuel cell power plant or an internal combustion engine.

The patented claimed set does not disclose that the fuel stream is diesel fuel. However, it would have been obvious to one having ordinary skill in the art at the time the invention was made to have modified the patented claimed process by treating a diesel fuel because any

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hydrocarbon fuel which contains sulfur can be treated in the process. Therefore, it would be expected that diesel would be effective to treat in the process the patented claimed process.

This is a <u>provisional</u> obviousness-type double patenting rejection because the conflicting claims have not in fact been patented.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

- 1. Determining the scope and contents of the prior art.
- 2. Ascertaining the differences between the prior art and the claims at issue.
- 3. Resolving the level of ordinary skill in the pertinent art.
- 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later

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invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

Claims 1-9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Setzer et al. (3,485,746).

Applicants are claiming several methods for desulfurizing a feed which involve processing an oxygenate containing feed over a nickel desulfurization station.

The reference of Setzer et al. (3,485,746) discloses a process for desulfurizing a hydrocarbon fuel containing organic sulfur compounds such as thiophenes and mercaptans for use in a fuel cell. See column 1, lines 20-25 and column 2, lines 35-43. The disclosed process is suitable for processes that are affected by trace amounts of sulfur. See column 1, lines 22-23. The process involves adding water (steam) to a fuel and contacting the water containing fuel with nickel metal. See Fig.1, page 1, column 1, lines 62-66. The desulfurization is conducted at a temperature of 500-900°F. See column 1, lines 69-70. The nickel bed is converted to nickel sulfide. See column 3, lines 26-29. The reference further teaches that it is thought that oxygen from the steam forms a protective layer on the nickel particles, thereby preventing undesirable coke formation in the bed. See column 3, lines 32-45.

The reference of Setzer et al.(3,485,746) succeeds in disclosing a process for desulfurizing a fuel suitable for use in fuel cells. The reference succeeds at disclosing the addition of an oxygenate in the form of water. In addition, the reference succeeds in disclosing a nickel reactant-absorbent for converting organic sulfur compounds to nickel sulfide which is considered to correspond to applicants' desulfurization station.

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Several differences are noted between the reference of Setzer et al.(3,485,746) and applicants' claimed invention. The reference is silent about the process effluent containing less than 0.05 ppm sulfur. The reference also does not disclose the fuel stream is a gasoline or a diesel fuel.

It would have been obvious to one of ordinary skill in the art at the time the invention was made to utilize a gasoline or diesel fuel because it appears that any <u>hydrocarbon fuel</u> can be employed in the process of Setzer to remove sulfurs. Therefore, one of skill in the art would use any hydrocarbon fuel including gasoline or diesel fuel which contains sulfur and it would be expected that the results would be the same or similar when using gasoline, diesel fuel, or any other hydrocarbon fuel in the process of Setzer et al. (3,485,746) because gasoline, diesel and the Setzer feed are a <u>hydrocarbon fuel</u>. Consequently, when gasoline or diesel are employed in the process of Setzer et al. (3,485,746), the gasoline or diesel product, which is well known, can be used in an internal combustion engine.

Since the modified process of Setzer is similar to the claimed process in terms of feedstock, reactant, and oxygenate, it would be expected that the product of Setzer would have less than 0.5 ppm sulfur as claimed.

Claims 10-12 rejected under 35 U.S.C. 103(a) as being unpatentable over Setzer et al. (3,485,746) in view of Jackson et al. (6,348,075).

See teachings of Setzer et al. (3,485,746) and statements of obviousness above.

Several additional differences are noted between the reference of Setzer et al.(3,485,746) and applicants' claimed invention.

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Setzer et al.(3,485,746) does not disclose maintaining the desulfurization station (nickel bed) at a temperature in the range of 300-450°F. The reference also does not disclose the production of isobutylene and methanol.

The reference of Jackson is cited to show that conventional gasoline contains oxygenates including, methanol, ethanol and ether. See column 8, line 38 through col. 9, line 3. Such compounds are known to have high blending octanes in an internal combustion engine.

It would have been obvious to one of ordinary skill in the art at the time the invention was made to treat the fuel containing oxygenates such as methanol, ethanol or MTBE (methyl tertiary butyl ether) as taught by Jackson et al. (6,348,075) because Jackson et al. (6,348,075) illustrates that it is conventional for such gasoline to contain oxygenates which are known to desirably have high blending octanes and because any hydrocarbon fuel can be used in the process regardless of whether or not the hydrocarbon fuel containing oxygenate (see the statement of obvious above). In addition, applicants' methanol/isobutylene production limitations are not considered to be patentable distinctions because the formation of isobutylene or methanol would naturally result from processing a feed containing the oxygenates of Jackson (6,348,075) over the nickel catalyst of Setzer et al. (3,485,746).

It would have been obvious to one of ordinary skill in the art at the time the invention was made to accomplish the desulfurization of Setzer et al. (3,485,746) at a temperature of 450°F because, from Fig. 2, one would learn that the process of Setzer et al. would be operable at a low temperature such as 450° C and at such temperature, it would be expected that the breakthrough time would be less. This may result in more frequent catalyst regenerating. However, operating at lower temperature would not affect the overall removal of sulfur in the process. Therefore,

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one of ordinary skill would be motivated to select a specific elevated temperature which accomplishes a desired level of desulfurization, such as the specific temperature claimed by applicants because it would be expected that the results would be the same or similar when operating the process of Setzer et al (3,485,746) at either 500 or 450° F.

Response to Remarks

The argument that the desulfurized fuel of Setzer et al. must have a sulfur content of less than about 20 ppm, since a sulfur content in the desulfurized bed which above the level indicates that bed is not functioning properly is not persuasive because the reference's disclosure of "amounts below those amounts detectable" encompasses applicants' 0.05 ppm. Applicants' sulfur amounts are considered to overlap those disclosed by Setzer et al. It is reminded that in an adsorption process, after an extended period of time in the process, the absorbability of the adsorbent is decreased. As a result, the amount of sulfur in the product would be increase and the adsorbent is needed to be regenerated (breakthrough). In the Setzer process, the breakthrough is considered when the fuel condensate was found at about 20 ppm of sulfur. This does not mean that during the process, the desulfurized fuel contains 20 ppm of sulfur (see col. 4, lines 34-40).

The argument that the examiner has not pointed out which parameters in Setzer et al. would have to manipulated in order to achieve an effluent having less than 0.05 ppm sulfur is not persuasive because the modification of Setzer is similar to the claimed process in terms of feedstock, reactant, and oxygenate. It would be expected that the product of Setzer would have less than 0.5 ppm sulfur as claimed.

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The argument that it is not obvious to modify the process of Setzer et al. to operate at a temperature of from 300 to 450° C is not persuasive. From Fig. 2, one would learn that the process of Setzer et al. would be operable at a low temperature such as 450° C and at such temperature, it would be expected that the breakthrough time would be less. This may result in more frequent catalyst regenerating. However, operating at lower temperature would not affect the overall removal of sulfur in the process. Therefore, the examiner maintains that one of ordinary skill would be motivated to select a specific elevated temperature which accomplishes a desired level of desulfurization, such as the specific temperature claimed by applicants because it would be expected that the results would be the same or similar when operating the process of Setzer at either 500 or 450° F.

The argument that it is extrapolated from Fig. 2 the breakthrough would be zero hours before reaching an operating temperature of 450° F is not persuasive because it is extrapolated that the breakthrough time would be about 4 hours.

Since the present Office Action has some different rejections than the co-pending application, arguments that are not relevant to the present Office Action will not be addressed.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Tam M. Nguyen whose telephone number is (571) 272-1452. The examiner can normally be reached on Monday through Thursday.

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Glenn Caldarola can be reached on (571) 272-1444. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Tam M. Nguyen Examiner

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TN